# Toxicity of Zinc on Growth of an Aquatic Macrophyte, *Ipomoea aquatica* Forsk

## LAITONJAM BEDABATI CHANU and ABHIK GUPTA\*

Department of Ecology and Environmental Science, Assam University, Silchar-788011, Assam, India.

http://dx.doi.org/10.12944/CWE.11.1.27

(Received: March 28, 2016; Accepted: April 15, 2016)

#### ABSTRACT

The effects of different concentrations of Zn on growth of an aquatic macrophyte, *Ipomoea aquatica* Forsk. were studied. Fresh weight, dry weight, shoot length, root length, number of nodes, development of leaves, and chlorophyll and carotene contents were the different growth parameters considered. Toxicity symptoms like browning and decaying of roots could be observed in plants treated at 22.7 mg L<sup>-1</sup> Zn as early as 3<sup>rd</sup> day of experiment while yellowing of older leaves appeared during the later period of exposure. High concentrations of Zn (12.71 – 22.7 mg L<sup>-1</sup>) significantly inhibited the growth of plant while lower Zn concentrations up to 4.09 mg L<sup>-1</sup> enhanced its growth. However, at 7.26 mg L<sup>-1</sup> Zn the chlorophyll as well as total carotene content in leaf of *I. aquatica* were significantly reduced from that in control on 5<sup>th</sup> day of exposure and subsequently the reduction was observed in lower concentrations. Thus, *I. aquatica* can be employed in biomonitoring of Zn polluted aquatic ecosystems using root browning, root and shoot growth inhibition, and chlorophyll and total carotene contents as sensitive biomarkers.

Keywords: Aquatic ecosystems, Biomonitoring, Carotene, Chlorophyll, Root browning.

#### INTRODUCTION

Natural environment, particularly the aquatic ecosystem is being disturbed by anthropogenic activities with rise in industrialization and urbanization. The decline in water quality of fresh water systems threatens its sustainability and has become one of the major environmental issues. The current scenario of Indian freshwater resources has been studied by many workers and appealed for their management as environmental problems. The increased concentrations of heavy metals in receiving water, especially lakes and rivers worsened the situation and threatened the biological components through bioaccumulation along the food chain and ultimately affect human being as well<sup>1-5</sup>.

Zn which is an essential micronutrient associated with metabolic activities in organisms can turn toxic at higher concentrations. Its concentration in various Indian aquatic ecosystems has reached alarming levels as reported by a number of scientific studies<sup>4,6-9</sup>. The main sources of Zn in fresh water ecosystems includes effluents from electroplating industries, smelting and refining, mining, paper industries, domestic sewage and agricultural runoff<sup>10</sup>. The immersion of idols and tazias is also one of the major reasons for increase level of Zn in lakes and rivers all over India<sup>11-13</sup>. Zn is one of the major constituents of paints for decorating both idols and tazias, and it has been reported that the number of idol immersion is being increased each year, which could increase load of heavy metals<sup>14</sup>.

Although Zn is a micronutrient, it becomes hazardous at high levels<sup>15</sup>. The growing concern in environmental aspects as well as the narrow window between its essentiality and toxicity<sup>16</sup> has generated interest in studying its effects on aquatic plants which play an important role in ecosystem functioning. In addition, a regular testing of Zn pollution in fresh water is required in the Indian context. Thus, the present study was carried out with the aim of exploring the potential of an aquatic macrophyte, Ipomoea aquatica Forsk. to be used in biomonitoring and phytotoxicity studies. This plant was selected on the basis of its geographical distribution, availability and adaptability. It is widely distributed not only in India, but in the entire South, South East, and East Asia<sup>17</sup> and well adapted to wide range of habitats. It is also reported to be grown in other parts of the world like Africa, Australia and United States of America (Austin, 2007)<sup>18</sup>. Further, it is perennial and mostly grows in moist soil, inundated floodplains, ditches, ponds, canals and sluggish rivers<sup>19</sup> and easy to cultivate due to its ability to proliferate by fragmentation, and produce adventitious roots and lateral branches that bear flowers and leaves, from its nodes20.

A number of studies have been conducted on Zn toxicity using aquatic plants. Growth inhibition was observed in *Eichhornia crassipes*<sup>21</sup>, *Lemna gibba*<sup>22</sup>, *Phragmites australis*<sup>23</sup>, *Iris psuedacarus*<sup>24</sup>, Spirodela polyrhiza<sup>25</sup> and Salvinia natans<sup>26</sup>. Another study reported the inhibition of growth induced by Zn stress in three aquatic macrophytes viz. Lemna minor, Elodea canadensis and Leptodictyum riparium, where L. riparium turned out to be the most resistant species with 50% growth inhibition at 100  $\mu$ M Zn<sup>27</sup>. In contrast to this, studies on Zn toxicity to A. aquatica is scarce. This study was, therefore, taken up to understand and assess the toxic effects of Zn on this aquatic macrophyte.

### MATERIALS AND METHODS

*Ipomoea aquatica* Forsk. was collected from uncontaminated ponds of Irongmara area in Cachar district, Assam, India, and washed with tap water. Stock cultures were grown following standard method<sup>28</sup>. The plants were grown in hydroponic tubs till new branches developed. These new branches were cut and planted in pots containing soil flooded with 50 % Hoagland nutrient media. The pH of nutrient media was adjusted at the range of 5.8-

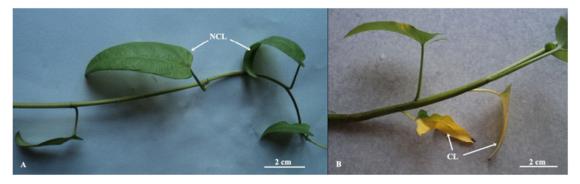


Fig.1: Zn induced change in appearance of leaf in *I. aquatica*; A- Non chlorotic leaf (NCL) of control; B- Chlorotic leaf (CL) of plant treated at 22.7 mg L<sup>-1</sup> Zn

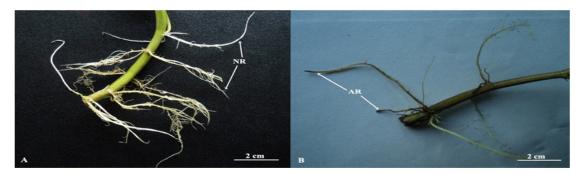


Fig. 2: Zn induced darkening of root in *I. aquatica*; A- Normal root (NR) of control; B- Affected root (AR) of plant treated at 22.7 mg L<sup>-1</sup> Zn

6.2. Healthy and fully grown shoots of similar shoot height were cut from the same mother plant, washed with tap water and acclimatized in 50 % Hoagland nutrient media for one week at 25-27°C, 12 h light with an intensity of 100-120 µmol<sup>-2</sup> s<sup>-1</sup> and 12 h dark periods. These acclimatized plants were exposed to half strength Hoagland nutrient media added with graded concentrations of Zn as ZnSO, 7H<sub>o</sub>O (actual Zn concentrations: 0.23 mg L<sup>-1</sup>, 2.27 mg L<sup>-1</sup>, 4.09 mg L<sup>-1</sup>, 7.26 mg L<sup>-1</sup>, 12.71 mg L<sup>-1</sup> and 22.7 mg L<sup>-1</sup>) for 15 days. Plants grown in 50 % Hoagland media without added Zn served as control. Water loss due to evaporation or transpiration was compensated by renewal of solutions every week. At the termination of the experiment on 15th day all control and Zntreated plants were removed, washed with distilled water, and softly blotted to remove excess water before fresh weight of the plants, their root length and lateral roots, and leaf area were measured. This was followed by drying in hot air oven at 60 °C till constant weight. Other growth parameters such as shoot length, new leaves, and number of nodes were measured at 3 day intervals. The percentage of primary roots bearing lateral roots (PRL %), and the ratio of lateral roots (LR) to the number of primary roots bearing lateral roots (PRL) was calculated. All these data were also obtained before the plants were exposed to Zn. Leaf area was measured by using ImageJ (http://imagej.nih.gov/ij/) software. All leaves were neatly clipped at their petioles and properly spread to take the image of the entire leaf area. Toxicity symptoms on leaf and root, such as darkening of roots and appearance of chlorosis

Table 1: Appearance of darkening of roots in *I. aquatica* on exposure to graded concentrations of Zn (n=5)

were noted during the experiment. For chlorophyll

Zn	Number	of plar	nts with	darken	ed roots
Conc. (mg L <sup>-1</sup> )	Day 3	Day 6	Day 9	Day 12	Day 15
0.23	0	0	0	1	1
2.27	0	0	0	1	1
4.09	0	1	2	2	2
7.26	0	1	2	3	3
12.71	1	2	2	4	5
22.7	3	5	5	5	5

estimation, fresh leaf was homogenized with 80 % acetone, centrifuged, and the absorbance of the supernatant was taken at 662, 645, and 470 nm for chlorophyll a (chl a), chlorophyll b (chl b) and total carotene, respectively, in a spectrophotometer. The concentrations of these pigments were calculated following standard formula<sup>[29]</sup> with the extraction solution used as blank.

Statistical significance of differences among the data sets was determined with One-way ANOVA, with multiple comparisons made by Tukey test. All tests were done using SPSS 20 software for Windows.

#### **RESULTS AND DISCUSSION**

Plants at all concentrations of Zn could survive the total exposure period of 15 days, although some toxicity symptoms were encountered in plants exposed to higher concentrations of Zn. Thus the survival pattern of the plant reflected its tolerance capacity up to Zn concentrations of 22.7 mg L<sup>-1</sup>. Among aquatic macrophytes, Typha agustifolia and Colocasia esculenta were found to be the most tolerant plants to heavy metals<sup>[30]</sup>. These two plants did not show any significant toxicity symptoms when they were grown in sediment contaminated with Zn concentration of 363 mg kg<sup>-1</sup>. Eichhornia crassipes when exposed to 20 mg L<sup>-1</sup> Zn did not show any morphological symptom of toxicity<sup>[31]</sup>. The present study revealed that plants treated at 22.7 mg L<sup>-1</sup> showed yellowing of older leaves (Fig. 1), while significant blackening of root tips occurred at 22.7 mg L<sup>-1</sup>Zn on the 3<sup>rd</sup> day of exposure (Fig. 2).

Blackening of root tips was also observed in plants treated at other concentrations with longer exposure time (**Table 1**). One way ANOVA showed significant reduction in length of PR at 7.26 - 22.7 mg L<sup>-1</sup> Zn (**Table 1**) which might be correlated to gradual decaying of roots along the length. Zn at 200 µg ml<sup>-1</sup> reduced root biomass in *Iris psuedacorus*<sup>24</sup>. However, symptoms like blackening and decaying of roots were not reported, although plants exposed to other heavy metals like Cd showed browning of root tips<sup>32</sup>. Dark root tips in *Vicia faba*<sup>33</sup> and cell death in root tissues of *Talinum triangulare*<sup>34</sup> were also reported due to oxidative stress induced by Pb accumulation. Thus, these symptoms of toxicity in roots can be used as a tool in biomonitoring of Zn pollution in water. The suppression of development of new PR led to a significant reduction in the PRL % as well as lateral roots in plants treated at 12.71 – 22.7 mg L<sup>-1</sup> Zn (**Table 2**). A significant reduction at p < 0.05 in LR: PRL was also observed in plants treated with 7.26 – 22.7 mg L<sup>-1</sup>Zn at the end of 15

day exposure as revealed by one way ANOVA and multiple comparisons by Tukey test (**Table** 2). Zn at concentrations of 0.23 and 2.27 mg L<sup>-1</sup>, on the other hand, enhanced growth of roots in *I. aquatica*.

Despite the toxic symptoms exhibited by *I. aquatica*, growth in terms of fresh weight and

Zn Conc. (mg L <sup>-1</sup> )	Increase in length of PR (cm)	No. of new PR	PRL %	LR:PRL
Control	23.77±2.43 <sup>ab</sup>	15.67±2.67ª	53.26±14.77 <sup>abc</sup>	5.81±1.33ª
0.23	39.23±9.68ª	16.33±1.33ª	68.88±6.20 <sup>ab</sup>	$5.00 \pm 0.25^{a}$
2.27	32±5.04 <sup>ab</sup>	19.33±2.40 <sup>ab</sup>	72.14±10.90 <sup>a</sup>	3.49±0.60 <sup>ab</sup>
4.09	20.7±2.92 <sup>b</sup>	24±1.15 <sup>b</sup>	40.09±11.25 <sup>abc</sup>	2.92±0.79 <sup>ab</sup>
7.26	3.33±4.08°	9.67±1.45°	38.15±5.08 <sup>bc</sup>	0.15±0.88 <sup>bc</sup>
12.71	-1.97±4.22°	5±1.73 <sup>cd</sup>	32.81±8.88°	-1.20±1.69°
22.7	-4.73±1.17°	1.67±1.20 <sup>d</sup>	-20.21±11.40 <sup>d</sup>	-1.92±1.05°

Table 2: Effects of Zn on root length, number of PR, PRL % and LR:PRL of *I. aquatica* at the end of 15 days

LR – Lateral roots; PR – Primary roots; PRL % - percentage of primary roots bearing lateral roots; Values are given as mean±SE; Values with different superscript letters in the column indicate significant differences at p < 0.05; '-'Decrease in each parameter with respect to the corresponding initial value

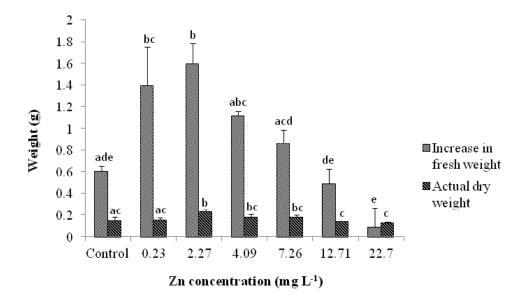


Fig. 3: Effects of Zn on fresh and dry weight of *I. aquatica* at the end of 15 days exposure; Values are given as mean $\pm$ SE; Values with different superscript letters indicate significant differences at p < 0.05

dry weight was not affected at Zn concentrations of 12.71-22.7 mg L<sup>-1</sup>, while 0.23-7.264.09 mg L<sup>-1</sup> Zn enhanced growth (**Fig. 3 & 4**). On the other hand, one way ANOVA with multiple comparisons by Tukey tests showed that Zn concentrations of 7.26 - 22.7 mg L<sup>-1</sup> had significant effect on number of new nodes (**Fig. 4**). There was also decline in increase in shoot length with significant effect at 22.7 mg L<sup>-1</sup> Zn as revealed by one way ANOVA at p < 0.05. Zn being a micronutrient probably enhanced photosynthesis and protein metabolism, thus increasing the growth of plants at low level of Zn<sup>[35]</sup>. Growth of *Sesuvium portulacastrum* was enhanced by Zn concentrations up to 300 mg kg<sup>-136</sup>. On the other hand, Zn at 2mM concentration could decrease shoot height substantially in *Phragmites* 

Zn Conc. (mg L <sup>-1</sup> )	No. of leaf fallen	No. of new leaf	Total leaf area (cm²)
Control	1.33±0.33ab	3±0 <sup>abd</sup>	29.30±4.3ª
0.23	0.33±0.33ª	$2\pm0^{ad}$	33.46±3.65ª
2.27	1.67±0.33 <sup>ab</sup>	2.67±0.33 <sup>abc</sup>	35.47±1.36ª
4.09	1.67±0.67 <sup>ab</sup>	3.67±0.33 <sup>bc</sup>	32.37±7.03ª
7.26	3±0.58 <sup>bc</sup>	2.67±0.88 <sup>cd</sup>	25.14±4.15ª
12.71	3.67±0.33°	2.33±0.33d	17.72±2.99ª
22.7	4.33±0.88°	0±0e	15.65±1.51 <sup>b</sup>

Table 3: Effects of Zn on leaf fall, new leaf and leaf area of *I. aquatica* at the end of 15 days

Values are given as mean $\pm$ SE; Values with different superscript letters in the column indicate significant differences at p < 0.05

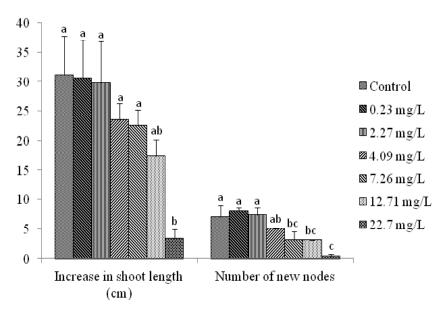


Fig. 4: Effects of Zn on increase in shoot length and number of nodes in *I. aquatica* at the end of 15 days exposure; Values are given as mean±SE; Values with different superscript letters indicate significant differences at p < 0.05

*australis*<sup>[23]</sup>. In another study, growth of *Hydrilla verticillata* was slightly affected by 0.1 and 1 mg L<sup>-1</sup> ZnO nanoparticle treatment for 3 weeks while 1000 mg L<sup>-1</sup> significantly reduced growth of the plant<sup>37</sup>.

Thus, there is a wide variation in response to heavy metals among plants<sup>38,39</sup>.

The results of the present study revealed that Zn at low concentrations induced new leaf

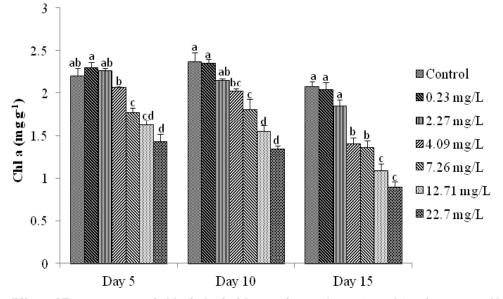


Fig. 5: Effect of Zn on content of chl a in leaf of *I. aquatica* on day 5, 10 and 15 of exposure; Values are given as mean $\pm$ SE; Values with different superscript letters indicate significant differences at p < 0.05

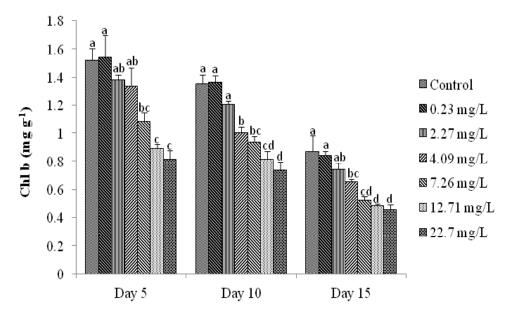


Fig. 6: Effect of Zn on content of chl b in leaf of *I. aquatica* on day 5, 10 and 15 of exposure; Values are given as mean $\pm$ SE; Values with different superscript letters indicate significant differences at p < 0.05

development as well as increased leaf area in *I. aquatica*, while at higher concentrations it affected leaf growth (**Table 3**). One way ANOVA showed significant differences in number of new leaf as well as leaf area among control and 22.7 mg L<sup>-1</sup> Zn treated plants at p < 0.05. Similar results were also reported in *Sesuvium portulacastrum* grown in soil amended with 100 – 600 mg kg<sup>-1</sup> Zn<sup>36</sup>. High concentrations of Zn also enhanced shedding of older leaf with significantly higher number of leaf fall at 12.71 and 22.7 mg L<sup>-1</sup>Zn from that in control. High Zn concentration resulted growth retardation and induces leaf senescence by disturbing key metabolic processes such as photosynthetic activity, pigment content and antioxidant systems<sup>40, 41</sup>.

The present study revealed that low Zn concentrations (0.23 and 2.27 mg L<sup>-1</sup>) enhanced the content of chl a in leaf of *I. aquatica* while it was significantly affected at 7.26 – 22.7 mg L<sup>-1</sup> Zn concentrations on 5<sup>th</sup> day of exposure as shown by one way ANOVA at p < 0.05. A dose and time – dependant reduction of chl a content was observed in subsequent days of exposure (**Fig. 5**). Similar results were also observed in case of chl b and

total carotene content (**Fig. 6** & **7**). Chlorophyll and total carotene content can be considered as effective biomarkers since significant decline was obtained at 7.26 mg L<sup>-1</sup> Zn after 5<sup>th</sup> day of exposure. In addition, symptoms like leaf yellowing was not observed at this concentration till the end of the experiment. Significant reduction in chlorophyll content with leaf yellowing was also observed in *Phragmites australis* treated with 1000 mg L<sup>-1</sup> ZnO nanoparticles<sup>37</sup>, although this dose is much higher than the dose of Zn which showed significant effect in the present study. Chl a, Chl b and carotenoid content in *Salvinia natans* was also significantly reduced by Zn treatment at the level of 10 mg L<sup>-126</sup>.

Thus, the results of this study revealed that *I. aquatica* can successfully be employed in toxicity studies in aquatic ecosystems polluted with Zn at a threshold limit of about 12 mg L<sup>-1</sup> by using root browning, reduction of growth in root and shoot and pigment content as tools for biomonitoring. This plant being resistant to Zn in terms of survival capacity coupled with its adaptive nature to wide range of habitats, has the prospect of being used in further studies like phytoremediation of Zn polluted areas.

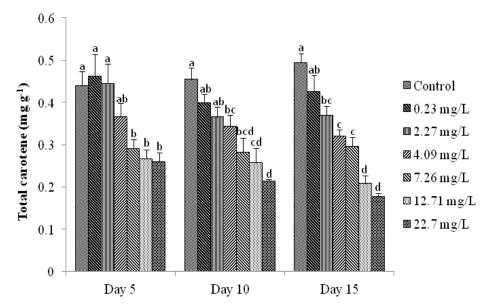


Fig. 7: Effect of Zn on content of total carotene in leaf of *I. aquatica* on day 5, 10 and 15 of exposure; Values are given as mean $\pm$ SE; Values with different superscript letters indicate significant differences at p < 0.05

#### ACKNOWLEDGEMENT

LBC would like to thank the University Grant Commission, New Delhi, India for financial

## REFERENCES

- Kumar, J.I.N., Soni, H. & Kumar, R.N., Biomonitoring of selected freshwater macrophytes to assess lake trace element contamination: a case study Nal Sarovar Bird Sanctuary, Gujarat, India. *Journal of Limnology*, 65(1): 9-16 (2006).
- Kaur, S. & Mehra, P., Assessment of heavy metals in summer & winter seasons in River Yamuna Segment flowing through Delhi, India. *Journal of Environment and Ecology*, 3(1): 149-165 (2012).
- Das, S., Mitra, A., Zaman, S., Pramanick, P., Chaudhuri, T.R. & Raha, A.T., Zinc, copper, lead and cadmium levels in edible finfishes from lower gangetic delta, *American Journal* of Bio-pharmacology Biochemistry and Life Sciences, 3(1): 8-19 (2014).
- Gupta, S.K., Chabukdhara, M., Kumar, P., Singh, J. & Bux, F., Evaluation of ecological risk of metal contamination in river Gomti, India: A biomonitoring approach, *Ecotoxicology and Environmental Safety*, **110**: 49-55 (2014).
- Bagul, V.R., Shinde, D.N., Chavan, R.P. & Patil, C.L., Causes and impacts of water pollution on rivers in Maharashtra- A review, *Research Journal of Chemical and Environmental Sciences*, 3(6): 1-4 (2015).
- Begum, A., Ramaiah, M., Harikrishna, Khan, I. & Veena, K., Heavy metal pollution and chemical profile of Cauvery River water. *E-Journal of Chemistry*, 6(1): 47-52 (2009).
- Subramanian, R., Nesakumari, C.S.A. & Thirunavukkarasu, N., Status of water quality and heavy metal pollution from Coovum River, Tamilnadu, India. Universal Journal of Environmental Research & Technology, 3(4): 483-489 (2013).
- Dhanakumar, S., Solaraj, G. & Mohanraj, R., Heavy metal partitioning in sediments and bioaccumulation in commercial fish species of three major reservoirs of river Cauvery delta region, India. *Ecotoxicology and Environmental Safety*, :145-151 (2015).

assistance in the form of UGC-BSR Fellowship during the course of the study.

- Anas, A., Jasmin, C., Sheeva, V.A., Gireeskumar, T.R. & Nair, S., Heavy metals pollution influence in the community structure of Cyanobacteria in nutrient rich Tropical Estuary. *Oceanography*, 3(1): 1000137, doi:10.4172/2332-2632.1000137 (2015).
- http://www.atsdr.cdc.gov/substances/ toxsubstance.asp?toxid=54/ (accessed on 29/02/2016).
- Reddy, M.V., Babu, K.S., Balaram, V. & Satyanarayan, M., Assessment of the effects of municipal sewage, immersed idols and boating on heavy metal and other elemental pollution of surface water of the eutrophic Hussainsagar Lake (Hyderabad, India). *Environmental Monitoring and Assessment*, 184: 1991-2000 (2012).
- Kaur, R., Effect of idol immersion on marine and fresh water-bodies. *Advances in Applied Science Research*, 3(4): 1905-1909 (2012).
- Malik, D., Singh, S., Thakur, J., Singh, R.K., Kaur, A. & Nijhawan, S., Heavy pollution of Yamuna river: An Introspection. *International Journal of Current Microbiology and Applied Sciences*, 3(10): 856-863 (2014).
- Aniruddhe, M. A case study of idol immersion in the context of Urban Lake Management. http://parisaraganapati.net/archives/83 (accessed on 24/02/2016).
- Nagajyoti, P.C., Lee, K.D. & Sreekanth, T.V.M., Heavy metals, occurrence and toxicity for plants: a review. *Environmental Chemistry Letters*, 8: 199-216 (2010).
- Dibofori-Orji, A.N. & Edori, O.S., Analysis of some heavy metals (Pb, Cd, Cr, Fe, Zn) in processed cassava flour (garri) sold along the road side of a busy highway, *Archives* of *Applied Science Research*, 7(2): 15-19 (2015).
- 17. Khumanleima Chanu, H. & Gupta, A., Necrosis as an adaptive response to copper toxicity in *Ipomoea aquatica* Forsk. and its possible application in phytoremediation.

*Acta Physiologiae Plantarum*, **36**; 3275-3281(2014).

- Austin, D.F., Water spinach (*Ipomoea aquatica*, Convolvulaceae) a food gone wild. *Ethnobotany Research & Applications*, 5: 123-146 (2007).
- Prasad, K.N., Shivamurthy, G.R. & Aradhya, S.M., *Ipomoea aquatica*, An underutilized green leafy vegetable: A review. *International Journal of Botany*, 4(1):123-129 (2008).
- Edi, H.H., & Ho, B.W.C., *Ipomoea aquatica* as a vegetable crop in Hong Kong. *Econmic Botany*, 23(1): 32-36 (1969).
- Hasan, S.H., Talat, M. & Rai, S., Sorption of cadmium and zinc from aqueous solutions by water hyacinth (*Eichhornia crassipes*). *Bioresource Technology*, **98**: 918-928 (2007).
- Megateli, S., Semsari, S. & Couderchet, M., Toxicity and removal of heavy metals (cadmium, copper and zinc) by *Lemna gibba*. *Ecotoxicology and Environmental Safety*, **72**(6): 1774-1780 (2009).
- Caldelas, C., Dong, S., Araus, J.L. & Weis, D.J., Zinc isotope fractionation in *Phragmites australis* in response to toxic levels of zinc. *Journal of Experimental Botany*, **62**(6): 2169-2178 (2011).
- Caldelas, C., Araus, J.L., Febrero, A & Bort, J., Accumulation and toxic effects of chromium and zinc in *Iris psuedacorus* L. *Acta Physiologiae Plantarum*, 34: 1217-1228 (2012).
- Hu, C., Liu, Y., Li, X. & Li, M., Biochemical responses of duckweed (*Spirodela polyrhiza*) to zinc oxide nanoparticles. *Archives of Environmental Contamination and Toxicology*, 64: 643-651 (2013).
- Hu, C., Liu, X., Li, X. & Zhao, Y., Evaluation of growth and biochemical indicators of *Salvinia natans* exposed to zinc oxide nanoparticles and zinc accumulation in plants. *Environmental Science and Pollution Research*, 21: 732-739 (2014).
- Basile, A., Sorbo, S., Conte, B., Cobianchi, R. C., Trinchella, F., Capasso, C. & Carginale, V., Toxicity, accumulation, and removal of heavy metals by three aquatic macrophytes. *International Journal of Phytoremediation*, (4): 374-387 (2012).

- Göthberg, A., Greger, M., Holm, K. & Bengtsson, B-E., Influence of nutrient levels on uptake and effects of mercury, cadmium, lead in Water spinach. *Journal* of Environmental Quality, **33**:1247-1255 (2004).
- Lichtenthaler, H.K. & Wellburn, A.R., Determination of total carotenoids and chlorophylls a and b of leaf in different solvents. *Biochemical Society Transactions*, 11; 591-592 (1983).
- Chayapan, P., Krautrachue, M., Meetam, M. & Pokethitiyook, P., Effects of amenmends on growth and uptake of Cd and Zn by wetland plants, *Typha augustifolia* and *Colocasia esculenta* from contaminated sediments. *International Journal of Phytoremediation*, 17: 900-906 (2015).
- Mishra, V.K. & Tripathi, B.D., Accumulation of chromium and zinc from aqueous solutions using water hyacinth (*Eichhornia crassipes*). *Journal of Hazardous Materials*, **164**(2-3): 1059-1063 (2009).
- Jamal, Q., Durani, P., Khan, K., Munir, S., Hussain, S., Munir, K. & Anees, M., Heavy metal accumulation and their toxic effects: Review. *Journal of Bio-Molecular Sciences*, 1(1-2): 27-36 (2013).
- Shahid, M., Pinelli, E., Pourrut, B., Silvestre, J. & Dumat, C., Lead-induced genotoxicity to *Vicia faba* L. roots in relation with metal cell uptake and initial speciation. *Ecotoxicology* and Environmental Safety, **74**(1): 78-84 (2011).
- Kumar, A., Prasad, M. N. V., Achary, V. M. M. & Panda, B. B., Elucidation of lead- induced oxidative stress in *Talinum triangulare* roots by analysis of antioxidant responses and DNA damage at cellular level. *Environmental Science and Pollution Research*, **20**: 4551– 4561(2013).
- Srivastava, S., Mishra, S., Dwivedi, S., Tripathi, R.D., Tandon, P.K. & Gupta, D.K., Evaluation of zinc accumulation potential of *Hydrilla verticillata. Biologia Plantarum*, **53**(4): 789-792 (2009).
- Kalaikandhan, R., Vijayarengan, P., Sivasankar, R. & Mathivanan, S., The effect of copper and zinc on the morphological parameters of *Sesuvium portulacastrum* L.

226

International Journal of Current Research and Academic Review, **2**(3): 105-120 (2014).

- Song, U. & Lee, S., Phytotoxicity and accumulation of zinc oxide nanoparticles on the aquatic plants *Hydrilla verticillata* and *Phragmites australis*: leaf-type-dependant responses. *Environmental Science and Pollution Research*, DOI 10.1007/s11356-015-5982-5 (2016).
- Yang, J., Tam, N.F-Y. & Ye, Z., Root porosity, radial oxygen loss and iron plaque on roots of wetland plants in relation to zinc tolerance and accumulation. *Plant Soil*, **374**: 815-828 (2014).
- Lin, Y-F. & Aarts, M.G.M., The molecular mechanism of zinc and cadmium stress response in plants. *Cellular and Molecular Life Sciences*, 69: 3187-3206 (2012).
- Nagajyoti, P.C., Lee, K.D. & Sreekanth, T.V.M., Heavy metals, occurrence and toxicity for plants: a review. *Environmental Chemistry Letters*, 8: 199-216 (2010).
- Gomes, M.P., Duarte, D.M., Carneiro, M.M.L.C., Barreto, L.C., Carvalho, M., Saores, A.M., Guilherme, L.R.G. & Garcia, Q.S., Zinc tolerance modulation in *Myracrodruom urundeuva* plants. *Plant Physiology and Biochemistry*, 67: 1-6 (2013).

227