Studies on Sediment Chemistry of River Yamuna with Special Reference to Industrial Effluents in Yamunanagar, India

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ABSTRACT

A study on sediment chemistry and water quality index of river Yamuna was conducted to understand the overall quality of river. Three sampling stations were selected: Station Y1 at the upstream of the river before the influx of effluents, Station Y2 at the point of influx and Station Y3 at 5 kilometres downstream from station Y2. The results showed the increasing values of pH, alkalinity, chloride and organic matter from station Y1 to Y2. Calculation of water quality index also categories station Y2 in bad or severely polluted zone. The correlation statistics showed the significant positive correlation between chloride and pH whereas significant negative correlation between organic matter and water quality index. The present work revealed the effect of Maskaranala’s effluents on the overall chemistry of sediments of river Yamuna.

Key words: Physicochemical characteristics, Pollution assessment, Sediment analysis, Water quality index.

INTRODUCTION

Global climate change and continues over load of pollution along with effluents from different industries have attracted various researchers to analyse the geochemical studies of river system (Torimanet et al., 2009, Gashi et al., 2009). India has ten major river systems among which Yamuna is the largest tributary river of Ganges in north India (Negiet et al., 1991). Yamunanagar (30° 6’ N latitude and 77° 17’ E longitude) is an important industrial city of Haryana. Along its path through Yamunanagar, Yamuna river get effluents from various industries like sugar mills, timber factories, paper industries etc. and sewage via maskaranala. In a river system, stream sediments have been widely used as environmental indicators and their chemical analysis can provide significant information for the assessment of anthropogenic activities and extent of pollution. Sediments also play an important role in the environmental studies of the rivers as they have long residence time for their interaction with the biotic components of the river’s ecosystem (Forstner and Wittmann, 1983). They play important role in nutrient cycle of aquatic environments and transport of essential nutrients as well as pollutants. Water quality index is an excellent management and general administrative tool in communicating water quality information and also plays important role in assessment of water resources for their suitability with reference to various uses (Chopra et al., 2014). In the present studies water quality indices of river Yamuna is correlated with its sediment chemistry. Some studies have been undertaken to assess the water quality of river Yamuna (Chopra et al., 2012, Bhatnagar et al., 2013) but studies dealing with sediment chemistry of river Yamuna with special reference to industrial effluents are very scanty. To reveal the effects of effluents on the sediment chemistry of river Yamuna, present studies have been conducted.
MATERIALS AND METHODS

Bearing in mind the pollution load, three sampling stations Y1, Y2 and Y3 were selected along the river stretch to perform practical aspects. Station Y1 is located near the village Kalanaur in district Yamunanagar at the upstream of the river, without any industrial discharge. This point is bathing and washing centre for the people of the village. Station Y2 is stationed 4-5 Kms downstream from station Y1. Here the effluents channels carrying industrial effluents via. maskaranala joins the river. Station Y3 point is stationed 5 Kms downstream from the station Y2 (Fig. 1).

Sediment samples were collected in polythene bags in triplicate on seasonal basis. Percentage moisture, electrical conductivity and pH were determined immediately, whereas for other physicochemical parameters i.e. soil alkalinity and chlorides, nitrate and organic matter, sediment was left to dry at air temperature and analysed according to standard procedures (Golterman et al., 1978; APHA, 1998) within the following 3 to 4 days. pH and conductivity were analyzed using MultiSet F Line three Water analysis kit (E Merck). Alkalinity, chlorides and organic matter were analysed by titrimetric method. Nitrate was determined spectrophotometrically (APHA, 1998). Brian Oram’s Water quality Index (WQI A) and Kaur’s Water Quality Index (WQI B) was calculated by using parameters such as pH, DO, BOD, Turbidity, alkalinity, calcium, magnesium, sulphate, chloride and nitrate of collected water samples from selected sites as per the standard references of Oram (2007) and Kaur et al. (2001).

RESULTS AND DISCUSSION

Moisture content depicted the water holding capacity of the sediments. During the present assessment of sediment quality, the maximum value of moisture was recorded during winter at stations Y2 and minimum during summer at Y1. The percentage of moisture significantly increased from station Y1 to Y2 and then decreased at station Y3(Table.1, Fig. 2). pH of the sediments was found to be alkaline throughout the study period. pH values increased from station Y1 to Y3. Seasonal fluctuations were well marked. Maximum values were recorded in post monsoon at station Y1 and Y3 while during summer at station Y2. Minimum values were during monsoon at station Y1 and Y3 and during winter at station Y2. Kaur et al. (1997), Tareqet al. (2013) also recorded high pH values during summer and low values during monsoon months. The values of conductivity varied between 200 to 215 µmhos cm⁻¹. The highest value of conductivity was recorded during summer at station Y1 and Y2 and during post monsoon at station Y3 whereas; lowest during monsoon at station Y1, post monsoon at station Y2 and summer at station Y3. Maximum value of conductivity in summer and minimum in monsoon was also recorded by Bath and Kaur (1999), Singh et al. (2007) and Singh et al. (2013). The values of conductivity decreased from station Y1 to Y2 and then increased at Y3.
Dilution of water during the rains caused a decrease in electrical conductance even of sediment. The mean values of alkalinity ranged between 3.3 mg g\(^{-1}\) to 4.1 mg g\(^{-1}\). Maximum alkalinity was recorded during summer at station Y2 and minimum during winter at station Y1. Similar trend were also recorded by Shastree et al. (1991) and Kaur et al. (1997). The values of alkalinity increased from station Y1 to Y2 and then decreased at station Y3 (Table 1, Fig. 2). Maximum value of chloride was recorded during post monsoon at station Y3 and minimum during monsoon at station Y1. Increase in the values during post monsoon may be due to decrease in water volume and increase in evaporation rate while decrease during monsoon may be due to dilution of water (Mandal and Das, 2011). Sewage water and industrial effluents are rich in chloride content and discharge of these waste waters results in greater chloride level in fresh waters (Guo-Qian and Niepin, 2000; Prabahar et al., 2012). The chloride content showed an increasing trend from station Y1 to Y2 and Y3 (Table 1, Fig. 2). The mean values of nitrate ranged between 0.2 mg g\(^{-1}\) (Y2) to 3.2 mg g\(^{-1}\) (Y3) during the study period. Nitrate showed a significant (P<0.05) decreasing trend from Y1 to Y2 and then increased at station Y3 (Table 1, Fig. 2). The maximum value was recorded during summer, monsoon and post monsoon at station Y3 while minimum during summer and winter at station Y2. Nitrate did not show any definite seasonal pattern. Organic matter increased from station Y1 to Y2 and

### Table 1: Mean values of sediment chemistry and water quality index (Mean ± S.E of mean) of river Yamuna at various stations

<table>
<thead>
<tr>
<th></th>
<th>Y1</th>
<th>Y2</th>
<th>Y3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Moisture%</td>
<td>45.4±0.36(^{C})</td>
<td>48.9±0.09(^{A})</td>
<td>46.9±0.12(^{B})</td>
</tr>
<tr>
<td>pH</td>
<td>7.3±0.03(^{B})</td>
<td>7.5±0.02(^{A})</td>
<td>7.5±0.02(^{A})</td>
</tr>
<tr>
<td>Conductivity µ m cm(^{-1})</td>
<td>203±2.94(^{B})</td>
<td>200±3.25(^{B})</td>
<td>215±2.76(^{A})</td>
</tr>
<tr>
<td>Alkalinity mg g(^{-1})</td>
<td>3.3±0.08(^{B})</td>
<td>4.1±0.09(^{A})</td>
<td>4.0±0.11(^{A})</td>
</tr>
<tr>
<td>Chloride mg g(^{-1})</td>
<td>2.2±0.12(^{B})</td>
<td>2.4±0.03(^{A})</td>
<td>2.4±0.07(^{A})</td>
</tr>
<tr>
<td>Nitrate mg g(^{-1})</td>
<td>1.4±0.04(^{B})</td>
<td>0.2±0.02(^{C})</td>
<td>3.2±0.06(^{A})</td>
</tr>
<tr>
<td>Organic matter %</td>
<td>2.8±0.09(^{B})</td>
<td>3.4±0.06(^{A})</td>
<td>3.1±0.10(^{AB})</td>
</tr>
<tr>
<td>WQI A</td>
<td>60±0.50(^{A})</td>
<td>56±0.50(^{A})</td>
<td>58±0.52(^{A})</td>
</tr>
<tr>
<td>WQI B</td>
<td>67.4±2.8(^{A})</td>
<td>39.9±4.3(^{B})</td>
<td>47.5±3.0(^{B})</td>
</tr>
</tbody>
</table>

Means with different letters in the same row are significantly (P<0.05) different. (Data were analyzed by Duncan's multiple range test)

### Table 2: Correlation between parameters of sediment chemistry and water quality indices

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Moisture</th>
<th>pH</th>
<th>Cond.</th>
<th>Alkalinity</th>
<th>Chloride</th>
<th>Nitrate</th>
<th>Organic matter</th>
<th>WQI A</th>
<th>WQI B</th>
</tr>
</thead>
<tbody>
<tr>
<td>Moisture</td>
<td>1</td>
<td>.822</td>
<td>-.269</td>
<td>.882</td>
<td>.822</td>
<td>-.471</td>
<td>.997</td>
<td>-.997</td>
<td>-.944</td>
</tr>
<tr>
<td>pH</td>
<td>-</td>
<td>1</td>
<td>.327</td>
<td>.933</td>
<td>1.0(^{**})</td>
<td>.115</td>
<td>.866</td>
<td>-.866</td>
<td>-.964</td>
</tr>
<tr>
<td>Cond.</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>.217</td>
<td>.327</td>
<td>.976</td>
<td>-.189</td>
<td>.189</td>
<td>-.063</td>
</tr>
<tr>
<td>Alkalinity</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>.993</td>
<td>.000</td>
<td>.918</td>
<td>-.918</td>
<td>-.988</td>
</tr>
<tr>
<td>Chloride</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>.115</td>
<td>.866</td>
<td>-.866</td>
<td>-.964</td>
</tr>
<tr>
<td>Nitrate</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>-.397</td>
<td>.397</td>
<td>.155</td>
</tr>
<tr>
<td>Organic</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>-1.0(^{**})</td>
<td>-.968</td>
</tr>
<tr>
<td>WQI A</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>.986</td>
</tr>
<tr>
<td>WQI B</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1</td>
<td>1</td>
</tr>
</tbody>
</table>

\(^{**}\) Correlation is significant at 0.01 level
then decreased at Y3 (Table 1, Fig. 3). Maximum
organic matter was recorded during winter at station
Y2. Generally high values of nutrients and organic
matter reflect a constant supply from sewage input
(Bath and Kaur, 1999). The increase in the values
of organic matter at Y2 may be due to influx of
effluents through maskaranala at station Y2 (Singh
et al., 2013).

According to Brian Oram’s water quality
index (WQI A) maximum value was recorded at
station Y1 and minimum at station Y2. Kaur’s water
quality index (WQI B) also showed the similar trend
of high value at Y1 and low at Y2. The present
investigation revealed that the water was of ‘bad
quality’ according to Brian Oram’s and severely
polluted as per the indexing of Kaur’s water quality
at station Y2 (Table 1, Fig. 3). The low values of
water quality index indicating that water was being
polluted at point of influx of effluents. The correlation
studies showed the significant positive correlation
between chloride and pH whereas significant
negative correlation between organic matter and
WQI A (Table 2). Correlation statistics revealed that
with increase in organic matter water quality of river
is deteriorating. High values of some
physicochemical parameters of sediments at station
Y1 and Y3 may be due to relatively more
anthropogenic activities at these stations.

Fig. 2: Seasonal variations in moisture content, pH, conductivity, alkalinity,
chloride and nitrate of sediments of river Yamuna at various stations
CONCLUSION

Study of different limnochemical parameters and organic matter of sediment samples revealed that the intensity of pollution increased as the river was subjected to sewage and industrial wastes. In the growing awareness of relationships between human health and water pollution, it is essential to undertake regular monitoring and surveillance of important aquatic ecosystems. In order to manage the pollution load of river Yamuna pass nearby Yamunanagar, it is recommended that various methods of sewage/industrial wastes treatment should be used before the disposal of effluents.

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